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Introduction

The question being addressed is: “**What is the best detector to use for non-aluminum metal and alloy identification, an SDD or a PIN detector?**” This paper shows that both a PIN and a SDD detector are equally adequate for identifying many common non-aluminum metals and alloys. The results section will focus the explanation on naval brass CD464 as an example material. Therefore, for this application **a PIN detector will work perfectly well and will be preferable to an SDD given the price difference.**

Comparing Detectors

The two main types of energy dispersive detectors are PINs and SDDs. The table below outlines a brief comparison between the two:

	Detection Area	Fe ⁵⁵ Resolution	Detector Inner Temp.	Upper Input Count Rate (ICR) Limit	Price
Typical SDD	10 - 50mm ²	120 - 160eV	-20 to -40°C	~500kcps	More expensive
Typical PIN	5 - 15mm ²	150 - 220eV	-20 to -40°C	~100kcps	Less expensive

Table 1: A brief comparison between typical SDD and PIN detectors.

A typical SDD has better performance over a PIN; they have a better ultimate energy resolution, and they are able to count more x-rays in a given time. The resolution is important in resolving x-ray events from different elements, and the counting is important for getting better statistics in a shorter time frame. A typical PIN detector has the major advantage of being less expensive, and is often used in XRF systems that are price sensitive.

PIN detectors are a good fit for many XRF applications, such as metal and alloy identification, which do not need the performance advantages that the SDD provide.

Experimental Conditions

The testing has a standard XRF set-up, shown in Figure 1. Moxtek’s 50kV, 4 Watt Ultra-Lite x-ray source with a tungsten anode and a 250µm beryllium window. The source was set at 50kV and at 15 or 20µA. The source-to-sample distance is 25mm, with a 70µm copper filter in front of the source. The sample-to-detector distance is 25mm for each detector. Both XPIN detectors have a 25µm thick beryllium window and SDD has a 12µm beryllium window. The signal from the detector was processed by Moxtek’s MXDPP-50. The X-ray source and the detectors have an aluminum sleeved brass collimator on them. The collimator is 11 mm long with a diameter of 3.8mm. The aluminum sleeve is necessary to eliminate the stray XRF signal from the brass, insuring the XRF signal is coming exclusively from the sample.

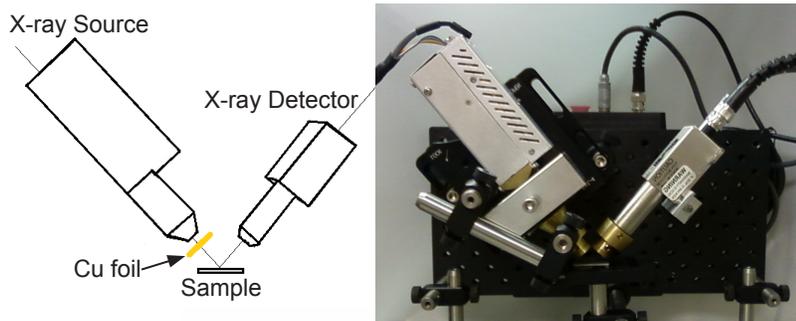


Figure 1. On the left is a sketch of the XRF setup, outlining the most critical parts. On the right is an image of the set up where all the components including collimators can be seen.

The Cu filter eliminates most of the x-rays from the source below ~15keV, giving a better signal-to-noise ratio in this region, but the Cu filter does let one tungsten L_{α} line through at ~8.3keV. The tungsten L_{α} line improves the excitation from Nickel and the lower Z elements, but also leads to a non-XRF peak which may confuse the untrained operator or an XRF algorithm. Figure 2 shows the XRF spectra from a clean plastic sample, comprising the Compton scattered tungsten L_{α} line and Compton scattered bremsstrahlung from the source. An SDD, XPIN6 and XPIN13 detector were compared for their XRF performance in identifying non-aluminum alloys and metals. Table 2 outlines the critical technical merits of each detector in this experiment.

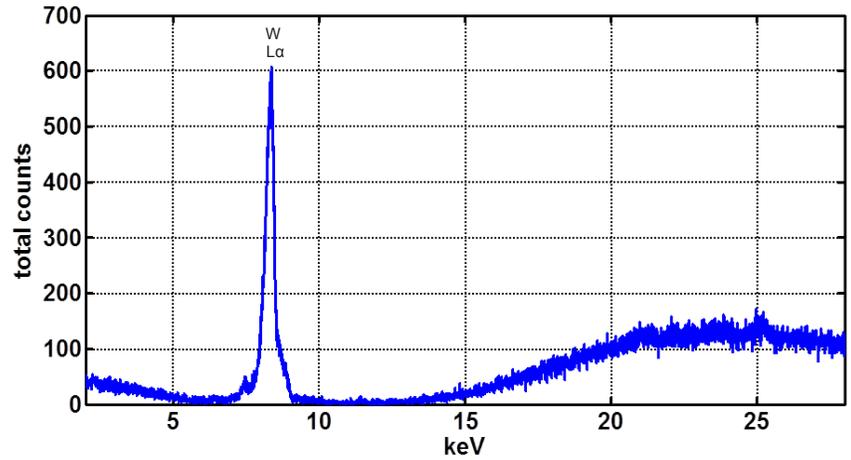


Figure 2. Spectra collected from a plastic sample, which shows the Compton scattered background from a clean XRF sample.

	Detector Area	Fe ⁵⁵ FWHM resolution	Naval Brass Spectra counts in 30 sec	Dead time	DPP Peaking time	Tube current	Detector Temp
SDD	20mm ²	150eV	487k	38%	8μsec	20μA	-45 °C
XPIN6	6mm ²	165eV	148k	21%	20μsec	20μA	-35 °C
XPIN13	13mm ²	200eV	205k	42%	20μsec	20μA	-35 °C

Table 2. A functional comparison between an SDD, XPIN6 and XPIN13 detector. Each detector was tested in a XRF setup (Figure 1) for non-aluminum metal alloy detection.

For this test, the X-ray tube was set at the same setting, resulting in different dead times on the detector. The SDD, as expected, has higher technical performance than the PIN detectors. The SDD has a lower Fe⁵⁵ FWHM resolution at a faster peaking time and more detecting area. This results in the SDD having roughly a 3X higher counting rate than the PIN detectors.



XRF Results

Naval brass is comprised of <0.1% Fe, 59-62% Cu, 39% Zn, 0.5-1.0% Sn and <0.2% Pb. Each detector recorded an XRF spectrum from a naval brass source for 30 seconds. Figure 3 shows the full spectrum from the SDD, XPIN6, and XPIN13, showing all the major elements. Figure 4 shows the same spectral data focused on the region from 5 to 11keV. One key point is all the element's K_{α} lines are separated well enough for clear identification. The SDD and PIN6 also have a distinction between the Zn K_{α} and Cu K_{β} . The XPIN13, which has lower resolution, does not have a clear distinction between these lines. Even so, in the end this line blurring did not affect the calculated elemental concentrations; therefore the XPIN13 still functioned well.

The collected spectra from the naval brass CD464 sample was run through an XRF fundamental parameters (FP) routine to turn the spectra into elemental concentrations. FP uses first principles; one inputs the tube settings, the detector characteristics and the spectrum; the algorithm then outputs elemental concentrations. Table 3 below outlines the physical parameters in the XRF set-up, which are needed as inputs into the FP routine.

Each detector, when properly set up in the FP program, gives nearly the same results. Table 4 below outlines the resulting concentrations from each detector. 30 seconds and 10 seconds XRF scans were taken on each detector. For comparison to the industry, scans of about 10 seconds or less is usual for handheld XRF instruments for metal identification. Comparing the 30 to 10 second scans show that the identification can easily be achieved for all the detectors in 10 seconds. The higher count rates of the SDD are not explicitly needed for sub-percent level element identification in short time frames of about 10 seconds for alloy and metal identification.

Using the FP routine, each detector gave an elemental concentration within 1% or less for each of the elements compared to each other detector. This level of accuracy is adequate for identifying naval brass.

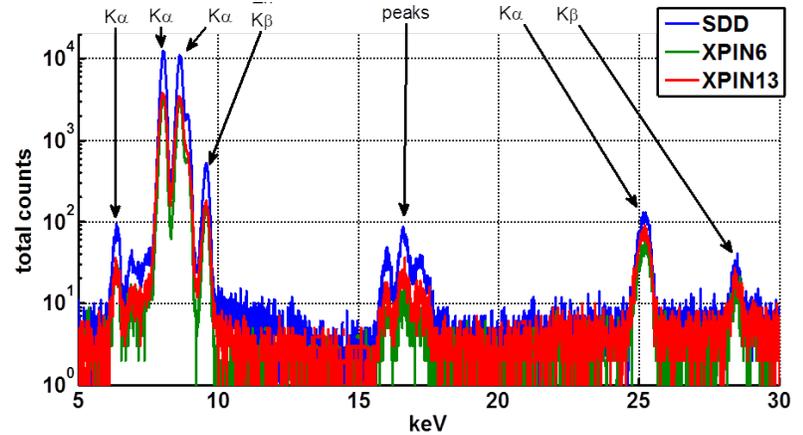


Figure 3. XRF Spectra collected from a naval brass sample, over 30 seconds, with all the major peaks labeled. Y axis of counts is in a log scale.

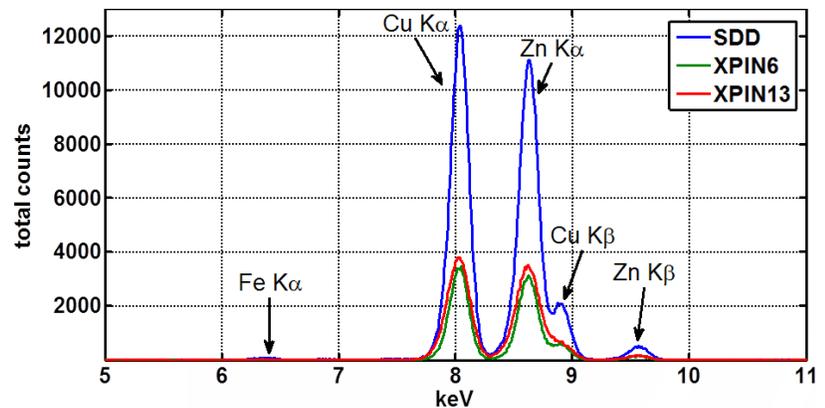


Figure 4. The same XRF Spectra collected in Figure 3, with a linear y-scale and energy range from 5 to 11keV.

Tube set-up	Tube Target	High Voltage	Beryllium Window Thickness	Tube- to- sample	Inc. angle/ take off angle	X-ray Source Filter
	W	49.3kV	250µm	25mm	90° / 90°	Cu, 75µm
Detector set-up	Detector Active Thickness	Detector Dead Layer	Beryllium Window Thickness	Sample- to- detector distance	Inc. angle/ Emer. angle	Detector Filter
SDD	500µm	0.15µm	12µm	25	135° / 45°	none
XPIN6	625µm	0.15µm	25µm	25	135° / 45°	none
XPIN13	625µm	0.15µm	25µm	25	135° / 45°	none

Table 3. The experiential inputs needed by the FP program to calculate the elemental concentration. The region of interest (ROI) of the FP program ranged from 2 to 40keV.

	Fe	Cu	Zn	Sn	Pb	Total counts in spectra
Tabulated Brass CD464	<0.1%	59-62%	b (39%)	0.5-1.0%	<0.2%	
SDD-30 sec	0.18	58.4	40.1	0.78	0.07	487k
PIN6-30 sec	0.20	58.8	39.6	0.99	0.05	148k
PIN13-30 sec	0.16	58.3	40.1	1.10	0.05	205k
SDD-10 sec	0.15	59.1	39.6	0.76	0.04	162k
PIN6-10 sec	0.17	58.6	39.8	1.08	0.05	54k
PIN13-10 sec	0.17	57.7	40.5	1.11	0.11	68k

Table 4. The resulting concentrations from the three compared detectors using the 30 second and 10 second scans from a Naval Brass CD464 sample. The last column gives the total number of x-ray events collected in each of the spectra.

Conclusion

Both a PIN and a SDD detector are equally adequate for identifying many common non-aluminum metals and alloys, such as naval brass CD464. Therefore **a PIN detector will work perfectly well and will be preferable to an SDD given the price difference for this application.**

